



Is the recommended pH for growing hardwood seedlings wrong?

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ARTICLE INFO

Citation:
 South DB (2019) Is the recommended pH for growing hardwood seedlings wrong? Reforesta 7: 81-108.
 DOI: <http://dx.doi.org/10.21750/REFOR.7.07.69>

Editor: Arnulfo Aldrete, Mexico
Received: 2019-01-17
Accepted: 2019-05-20
Published: 2019-06-28



Abstract

Two schools of thought address the optimum soil pH (measured in water) for growing hardwood seedlings in bareroot nurseries. One school uses nutrient surveys in non-fertilized forests to determine the best pH range for growing seedlings in fertilized nurseries. Some students of this school believe hardwood seedlings grow best at pH 6.0 to 7.5. In contrast, another school relies on research from pH trials to conclude that fertilized hardwoods can grow well in soils that range from pH 4.5 to 6.0. This article compiles some of the findings from seedbed and greenhouse trials and attempts to use data to dispel a few myths about the “optimum pH” for growing hardwood seedlings. Greenhouse trials suggest many fertilized hardwoods grow better in acid soils (pH 4-6) than in nearly neutral soils (pH 6.0-7.5). The optimal pH for growth differs among species and, therefore, it is a myth that all hardwood seedlings grow best at pH 6 to 7.5. Most nursery managers in the southern United States grow bareroot hardwoods between pH 4.8 and 6.0.

Keywords

Acidity; Bareroot; Micronutrient; Nursery; Nutrient deficiency

Contents

1	Introduction	82
2	Bareroot nurseries	85
3	Container nurseries	87
4	Irrigation pH	89
5	Genetic differences	88
5.1	<i>Carya</i>	91
5.2	<i>Ilex</i>	91
5.3	<i>Liquidambar</i>	91
5.4	<i>Paulownia</i>	91
5.5	<i>Populus</i>	92
5.6	<i>Quercus</i>	92
6	Warnings about low soil pH	93
7	Advantages of low pH	93
8	Sulfur	94
9	Lime	95
10	Problems with some conclusions	95
11	Future research	96
12	Conclusions	97
13	Acknowledgments	97
14	References	97

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1 Introduction

There is no consensus on the recommended pH range (measured in water) for growing hardwood seedlings in nurseries. In some cases, recommended ranges do not even overlap (Table 1). Although many recommendations involve a minimum pH of 5.5 or 6.0, some managers ignore this recommendation and grow bareroot hardwood seedlings in soils at pH 4.8 (Table 2). A pH range of 4.5 to 6.0 is also suitable for establishing plantations of various species (see Table 3 for scientific names). Sometimes hardwoods are established on acid strip-mined sites to restore these areas.

Table 1. The recommended soil pH for bareroot and container-grown hardwoods varies considerably. Several authors suggest seedbeds have a minimum pH (water) of 5.5 or greater.

Crop	Nursery	pH range	Reference
hardwoods	bareroot	6.5-7.5	Youngberg 1984
hardwoods	bareroot	6.5-7.0	Leaf et al. 1978
<i>Fraxinus</i>	bareroot	6.2	Wilde and Patzer 1940
hardwoods	bareroot	6.0-8.0	Laurie and Chadwick 1931
hardwoods	bareroot	6.0-8.0	Engstrom and Stoeckler 1941
hardwoods	bareroot	6.0-7.0	Solan et al. 1979
hardwoods	bareroot	6.0-7.0	Edwards 1986
hardwoods	container	6.0-7.0	Landis 1989
hardwoods	bareroot	6.0-8.0	Meyer 1993
hardwoods	bareroot	6.0-7.0	Briggs 2008
hardwoods	bareroot	6.0	Bonner and Broadfoot 1964
hardwoods	bareroot	6.0	Peevy 1976
hardwoods	bareroot	6.0	Williams and Hanks 1976
hardwoods	bareroot	6.0	Cordell and Filer 1984
hardwoods	bareroot	5.8-6.5	Steinbeck and May 1977
hardwoods	bareroot	5.8-6.2	Pritchett and Fisher 1987
<i>Acer, Tilia, Ulmus</i>	bareroot	5.8	Wilde and Patzer 1940
hardwoods	bareroot	5.6-6.8	Davey 1973
hardwoods	bareroot	5.6-6.8	Stone 1980
<i>Acer, Carya, Juglans</i>	bareroot	5.5-7.3	Wilde 1958
hardwoods	bareroot	5.5-6.5	Stoeckeler and Jones 1957
hardwoods	bareroot	5.5-6.5	Turner 1970
most hardwoods	container	5.5-6.5	Dumroese et al. 2012
hardwoods	container	5.5-6.5	Wilkinson et al. 2014
hardwoods and pines	bareroot	5.5-6.5	Landis 2008
<i>Populus and Fraxinus</i>	bareroot	5.5-6.5	Aldhous and Mason 1994
hardwoods	container	5.4-6.0	Mathers et al. 2007
<i>Betula</i>	bareroot	5.3	Wilde and Patzer 1940
<i>Populus</i>	bareroot	5.0-6.5	Jobling 1990
<i>Prunus, Quercus</i>	bareroot	5.0-6.0	Wilde 1958
<i>Quercus</i>	bareroot	5.0-6.0	Storandt 2002
most tree species	bareroot	5.0-6.0	Morby 1984
hardwoods	bareroot	5.0-6.0	Davey and McNabb 2019
<i>Acer</i>	bareroot	5.0-5.6	Altland 2006
hardwoods	bareroot	5.0-5.5	Aldhous and Mason 1994
<i>Ilex, Magnolia</i>	Bareroot	4.5-6.0	Laurie and Chadwick 1931
woody species	container	4.5-5.5	Whitcomb 2009
acid loving hardwoods	container	4.0-5.0	Dumroese et al. 2012

One might ask why strip-mined soils (with pH <5) are suitable for planting various hardwood species (Davis et al. 2012) while others warn against growing hardwoods in a fertilized nursery soil with a pH less than 5.5. Although seedling survival is low when strip-mined soils have a pH less than 3.5 (Davidson 1979; Hensley and Carpenter 1986), many planted hardwoods grow well in non-fertilized soils with pH 4 to 7.5 (Table 3, Ashby et al. 1989). Unlike most reclamation sites, fertilizers and irrigation are applied to increase growth of hardwoods in nurseries and several authorities indicate that fertilized hardwoods can grow well over a wide pH range. For example, yellow poplar and sycamore grow well on both alkaline (pH 7.3-8.0) and acid (pH 4.6) soils (Wilde 1954). This agrees with Peevy (1976) who said that hardwood seedlings “may be grown over a wide range [of] pH [if] the proper concentrations of essential mineral elements are maintained.” In contrast, pines have a narrower pH range than hardwoods (Benzian 1965; Mizell 1980; South 2017).

After reviewing the literature (Table 1), it became apparent that the pH 5.0-5.5 recommendations by Aldhous and Mason (1994) were based on empirical nursery trials (Benzian 1965) while none of the higher recommendations (minimum = pH 6.0 or 6.5) were based on pH trials from hardwood seedbeds. Instead, they were based on various assumptions about pH ranges observed in forests (e.g. Burns and Honkala 1990; DeVall 1943; Spurway 1941) and on assumptions about nutrient availabilities in non-fertilized agronomic soils (Brady 1974).

Table 2. The pH (water) of operational 1-0 hardwood seedbeds at time of sowing.

State	Year	Lowest pH	Highest pH
Alabama	2018	5.5	5.8
Arkansas 1	2015	5.2	5.8
Arkansas 2	2018	4.7	5.0
Georgia	2014	4.8	5.8
Indiana	2017	5.6	6.4
Mississippi	1981	4.9	4.9
North Carolina 1	1984	5.5	5.5
North Carolina 2	2016	4.8	4.8
South Carolina	2017	5.6	5.6
Tennessee	2017	4.9	5.5
Virginia 1	1999	5.2	6.2
Virginia 2	2016	5.2	5.7

Table 3. Estimates of “preferred” and “acceptable” soil pH (water) for plantations of selected hardwoods.

Common name	Species	“Preferred” pH range	Acceptable pH range	Reference
Black walnut	<i>Juglans nigra</i> L.	6.0-8.0	--	
Fremont cottonwood	<i>Populus fremontii</i> S. Watson	6.0-8.0	--	
Choke cherry	<i>Prunus virginiana</i> L.	6.0-8.0	--	
Basswood	<i>Tilia glabra</i> L.	6.0-7.5	--	
American elm	<i>Ulmus americana</i> L.	6.0-7.5	--	
Black locust	<i>Robinia pseudoacacia</i> L.	6.0-7.5	--	Spurway 1941
Sweetgum	<i>Liquidambar styraciflua</i> L.	6.0-7.0	--	
Yellow poplar	<i>Liriodendron tulipifera</i> L.	6.0-7.0	--	
Eastern redbud	<i>Cercis canadensis</i> L.	5.5-6.5	--	
Dogwood	<i>Cornus florida</i> L.	5.0-7.0	--	

Continuation of Table 3.

Common name	Species	"Preferred" pH range	Acceptable pH range	Reference	
White oak	<i>Quercus alba</i> L.	5.0-6.5	--		
Japanese holly	<i>Ilex crenata</i> Thunb.	5.0-6.0	--		
Sweet birch	<i>Betula lenta</i> L.	4.5-7.0	--		
Sweetbay magnolia	<i>Magnolia virginiana</i> L.	4.0-5.0	--		
Quaking aspen	<i>Populus tremuloides</i> Michx.	3.8-5.5	--		
Cottonwood	<i>Populus deltoides</i> Bartr. ex Marsh.	5.5-7.5	4.5-8.5		
Pecan	<i>Carya illinoensis</i> (Wangenh.) K. Koch	5.5-7.5	4.5-8.5		
Green ash	<i>Fraxinus pennsylvanica</i> Marsh.	5.5-7.5	4.5-8.5		
Sweetgum	<i>Liquidambar styraciflua</i> L.	5.5-7.5	4.5-8.5		
Sycamore	<i>Platanus occidentalis</i> L.	5.5-7.5	4.5-8.5		
Yellow poplar	<i>Liriodendron tulipifera</i> L.	4.5-7.5	4.5-8.5		
Cherrybark oak	<i>Quercus falcata</i> var. <i>pagodaefolia</i> Ell.	4.5-5.5	4.0-7.5	Baker and Broadfoot 1979	
Nuttall oak	<i>Quercus nuttallii</i> Palmer	4.5-5.5	4.0-7.5		
Shumard oak	<i>Quercus shumardii</i> Buckl.	4.5-5.5	4.0-7.5		
Swamp chestnut	<i>Quercus michauxii</i> Nutt.	4.5-5.5	4.0-7.5		
Water oak	<i>Quercus nigra</i> L.	4.5-5.5	4.0-7.5		
Willow oak	<i>Quercus phellos</i> (Wangenh.) K. Koch	4.5-5.5	4.0-7.5		
Hackberry	<i>Celtis occidentalis</i> L.	4.0-8.5	4.0-8.5		
Sugarberry	<i>Celtis laevigata</i> Wild.	4.0-8.5	4.0-8.5		
Princess tree	<i>Paulownia tomentosa</i> Siebold & Zucc.	6.0-8.0	--		
Live oak	<i>Quercus virginiana</i> Mill.	6.0-7.0	--		
Dogwood	<i>Cornus florida</i> L.	5.0-8.0	--		
Black walnut	<i>Juglans nigra</i> L.	5.0-7.5	--		
American beech	<i>Fagus grandifolia</i> Ehrh.	5.0-7.5	--		
Yellow poplar	<i>Liriodendron tulipifera</i> L.	5.0-7.0	--		
Yellow birch	<i>Betula alleghaniensis</i> Britton	5.0-6.5	--		
Pecan	<i>Carya illinoensis</i> (Wangenh.) K. Koch	4.8-7.5	--		
Blackgum	<i>Nyssa sylvatica</i> Marshall	4.6-7.0	--		
Black locust	<i>Robinia pseudoacacia</i> L.	4.6-8.2	--		
White oak	<i>Quercus alba</i> L.	4.5-6.2	--	Londo et al. 2006	
Northern red oak	<i>Quercus rubra</i> L.	4.5-6.0	--		
Sycamore	<i>Platanus occidentalis</i> L.	4.4-7.5	--		
Red maple	<i>Acer rubrum</i> L.	4.4-7.5	--		
Green ash	<i>Fraxinus pennsylvanica</i> Marsh.	3.6-7.5	--		
Cottonwood	<i>Populus deltoides</i> Bartr. ex Marsh.	3.6-7.5	--		
Sweetgum	<i>Liquidambar styraciflua</i> L.	3.6-7.5	--		
Nuttall oak	<i>Quercus nuttallii</i> Palmer	3.6-6.8	--		
Willow oak	<i>Quercus phellos</i> L.	3.6-6.3	--		
Water oak	<i>Quercus nigra</i> L.	3.6-6.3	--		
Black cherry	<i>Prunus serotina</i> Ehrh.	3.0-5.0	--		
Red alder	<i>Alnus rubra</i> L.	--	--		
Balsam poplar	<i>Populus balsamifera</i> L.	--	--		
Sawtooth oak	<i>Quercus acutissima</i> Carruth.	--	--		

2 Bareroot nurseries

Blanche Benzian (1965) conducted 16 pH trials with hardwoods at UK Forestry Commission nurseries (1948-1954). When lime (CaCO_3) was added to pH 4.2 seedbeds at the Wareham Nursery, some species grew taller while other were shorter (Table 4). Soon after, Aldhous (1972) summarized Benzian's research: "There has been little research into the pH responses of hardwoods. However, practical experience and limited research show that most hardwoods, e.g. oak, alder, beech, birch and [*Acer*] grow well on acid soils, but there are some important exceptions, namely poplars and ash which require slightly acid or neutral soils..."

Table 4. The effect of adding lime (CaCO_3) and aluminum sulfate (335 kg ha^{-1}) on soil pH (water) and on seedling height (mm) at the Wareham Nursery in the UK (Benzian 1965).

Species	Year	Untreated	Lime	Additional lime	Aluminum sulfate
	1948	pH 4.2	pH 5.3	pH 5.8	pH 3.8
<i>Betula lutea</i> L.	1948	164	121	87	61
<i>Fagus sylvatica</i> L.	1948	86	113	120	60
<i>Quercus robur</i> L.	1948	75	96	81	65
	1949	pH 4.0	pH 5.7	pH 6.5	pH 4.0
<i>Betula</i> spp.	1949	680	313	246	crop failed*
<i>Fraxinus excelsior</i> L.	1949	20	29	26	crop failed*

* Whitcomb (2018) recommends that aluminum sulfate not be used to lower soil pH.

Only a few trials have been conducted with lime in hardwood seedbeds in the United States. Lime was applied before sowing hardwood seedbeds at the Edwards Nursery (Morganton, NC) and Union Camp Nursery at Murfreesboro, NC (Deines 1973). At the Edwards Nursery in October, check plots and the high-lime plots ($1,121 \text{ kg ha}^{-1}$) averaged pH 5.9 and 6.8, respectively. The high lime treatment reduced growth of sweetgum (Figure 1) and a northern seed source of sycamore.

Crannell et al. (1994) conducted trials at bareroot nurseries in Oregon (pH 5.6) and Washington (pH 6.6). At the nursery in Washington, temporarily increasing acidity to pH 5.9 increased the production of red alder which increased nursery profits. In contrast, increasing soil pH to 7.2 with CaO had no effect on seedling biomass at the Oregon nursery.

While lime increases soil pH, some fertilizers (e.g. ammonium nitrate and ammonium sulfate) will lower soil pH. Trials at the Union Camp Nursery in Virginia indicated that sweetgum and green ash grew best when fertilized with ammonium sulfate which reduced soil pH to 4.5 (Figure 2; Davey 1996; Stone 1980; Villarrubia 1980). Perhaps due to these data, Davey (1996) did not mention what pH range was "optimum" for growing bareroot seedlings. Once the word spread that hardwood seedlings could be produced in soils with pH less than 5.0, several nursery managers abandoned the practice of using different soil pH targets for pines and hardwoods. In nurseries, both pines and hardwoods can be grown at the same pH range (Landis 2008; Morby 1984; Tinus 1979; Williams 1972). Good quality sweetgum, sycamore, green ash, yellow poplar, white oak and Nuttall oak have been grown in seedbeds that range from pH 4.8 to 6.0 (dos Santos 2006; Lamar and Davey 1988; Marx 1979; Rentz 1996). Whitcomb (2009) says that most woody plants grow best at soil pH 4.5 to 5.5.

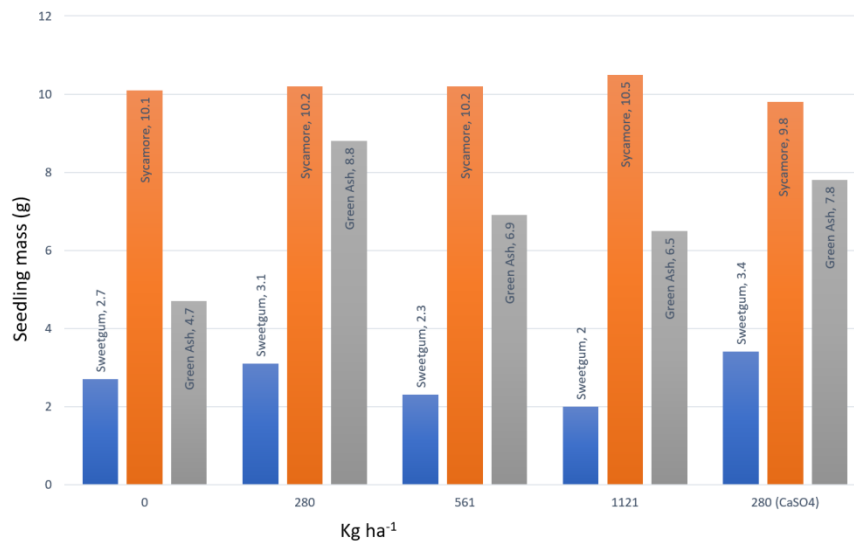


Figure 1. Data from North Carolina (Deines 1973) illustrate the effect of lime (CaCO₃) on dry mass of sweetgum, sycamore (Edwards Nursery at Morganton) and green ash (Union Camp Nursery at Murfreesboro) seedlings. Standard errors are 0.2, 0.8 and 1.5 g for sweetgum, sycamore and green ash respectively. Although the high rate of lime (pH 6.8 at lifting) reduced shoot mass and root-collar diameter of sweetgum seedlings, a significant increase occurred when 280 kg of gypsum (CaSO₄; pH 5.7) was applied to sweetgum seedbeds. Neither the lime nor CaSO₄ treatments had a significant effect ($\alpha=0.05$) on green ash seedlings. For sycamore, adding lime increased growth of a southern seed source but decreased growth of a northern seed source (data not shown).

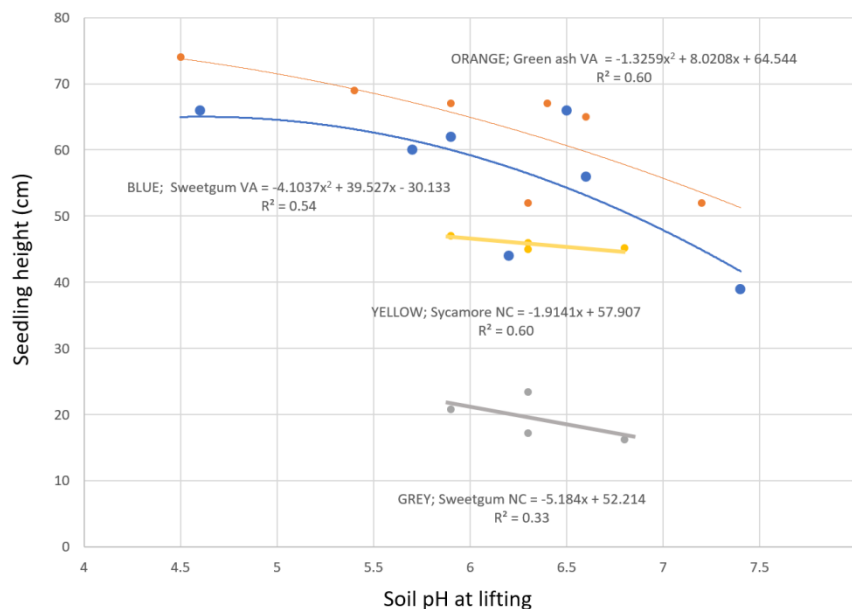


Figure 2. The correlations between soil pH (water) and height growth of bareroot hardwood seedlings. Data are from trials at the Edwards Nursery in North Carolina (Deines 1973) and from the Union Camp Nursery at Capron, Virginia (Villarrubia 1980).

3 Container nurseries

Sometimes problems with disease and nutrition occur when seedlings are grown in containers filled with 100% soil (Böhlenius et al. 2016; Chen et al. 2006; Hendrickson and Veihmeyer 1941; Wilkinson et al. 2014). Therefore, most hardwood seedlings are grown in containers using media composed of peat, perlite and vermiculite (Dumroese et al. 2012; Enebak 2011; Landis 1989). It appears that when compared to inorganic soil, the optimum pH range is greater when hardwoods are grown in a soil-less potting-mix. For example, height growth of birch grown in containers was not related to pH that ranged from pH 3.8 to 7 (Rikala and Jozefek 1990), while heights of bareroot birch seedlings were negatively related to soil pH (Table 4).

In Oklahoma, container-grown, woody ornamentals grew well at pH 3.0 when seedlings were fertilized with calcium (Ca) and magnesium (Mg), (Whitcomb 1983). "Only under conditions of high concentration of calcium, magnesium, sodium or bicarbonates of these elements does pH begin to influence micronutrient nutrition of plants in containers." After years of "frustration with pH and what it means or doesn't mean" Whitcomb (1988) now believes pH has little to do with the nutrition of growing seedlings in containers.

While media pH may not be a major factor affecting growth of properly fertilized seedlings in containers (Whitcomb 1988), various studies indicate biomass of several hardwoods is greater when media pH ranges from pH 4 to pH 5.5 (Table 5, Handreck 1989). Results from these trials can be used to reject the hypothesis that container-grown hardwoods grow best at pH 6-8. In one study (Beyer et al. 2013), oak seedlings grown at pH 7.4 were slightly chlorotic and were smaller than seedlings grown in pH 3.3 soil (Figure 3).



Figure 3. The effect of lime (reagent grade powdered amorphous CaCO_3 plus MgCO_3) on growth (11-weeks) of northern red oak seedlings in a greenhouse. Soil on the left was not limed (pH 3.3, $\text{Ca} = 42 \mu\text{g g}^{-1}$, $\text{Mg} = 11 \mu\text{g g}^{-1}$, $\text{Al} = 44 \mu\text{g g}^{-1}$) and was not fertilized while the pot on the right (pH 7.4, $\text{Ca} = 590 \mu\text{g g}^{-1}$, $\text{Mg} = 150 \mu\text{g g}^{-1}$, $\text{Al} = 0.2 \mu\text{g g}^{-1}$) contained soil that was treated with 50 g of lime per liter (Beyer et al. 2013). The lime treatment had no significant effect on root mass (left = 9.4 g; right = 9.6 g) but leaf mass (left = 8 g; right = 6 g) was reduced by the lime treatment. Better growth in acid soil (pH 3.3 – 5.4) casts doubt on the belief that the optimum pH range for growing hardwood seedlings is pH 6.0 to 8.0 (Engstrom and Stoeckler 1941, Laurie and Chadwick 1931, Meyer 1993). The soil was from the Hazleton series: loamy-skeletal, siliceous, active, mesic Typic Dystrudept. Foliar concentrations were: unfertilized = $\text{Ca} 0.5\%$, $\text{Mn} 690 \mu\text{g g}^{-1}$, $\text{Cu} 7.5 \mu\text{g g}^{-1}$, $\text{Zn} 110 \mu\text{g g}^{-1}$; limed = $\text{Ca} 1.8\%$, $\text{Mn} 32 \mu\text{g g}^{-1}$, $\text{Cu} 2.8 \mu\text{g g}^{-1}$, $\text{Zn} 19 \mu\text{g g}^{-1}$. (Photo by Nelson Beyer, United States Geological Survey 2009).

Table 5. Examples of greenhouse trials demonstrating the change (%) in plant mass (mg) or foliage (f) mass when hardwoods are grown at varying pH (water) levels. In most cases, mass increased when plants were grown at pH 4 to 5.5.

Genus	Comment	pH #1	pH #2	Mass for pH#1 (mg)	Mass for pH#2 (mg)	Change in mass with decreased pH (%)	Reference
<i>Liquidambar</i>	GM endo	6.7	5.1	2,550	610	-76	Davis et al. 1993
<i>Carya</i>		6.1	4.9	108,000	35,000	-67	Johnson and Hagler 1955
<i>Paulownia</i>		6.0	5.0	2,600	1,280	-51	Melhuish et al. 1990
<i>Quercus</i>		6.8	4.8	14,400	7,700	-46	Crews and Dick 1998
<i>Trichilia</i>		6.3	4.5	310	210	-32	Ramlall et al. 2015
<i>Betula</i>		7.0	5.0	30,800	23,500	-24	Zhang et al. 2016
<i>Leucaena</i>	GA endo	6.0	5.0	8,400	6,900	-18	Soedarjo and Habte 1995
<i>Eucalyptus</i>		5.2	4.0	18,400	16,200	-12	Gabriel et al. 2018
<i>Populus</i>	family 1	6.0	5.0	9,100	8,500	-7	DesRochers et al. 2003
<i>Alnus</i>		5.4	4.7	1,810	1,730	-4	Granhall et al. 1983
<i>Populus</i>	hybrid	7.2	5.8	9,000	8,800	-2 (f)	Timmer 1985
<i>Cornus</i>		7.0	5.0	12,400	12,500	+1	Zhang et al. 2016
<i>Alnus</i>		5.4	4.7	980	1000	+2	Granhall et al. 1983
<i>Populus</i>		5.7	4.6	14,500	14,800	+2	Hjelm and Rytter 2016
<i>Betula</i>		6.0	4.0	2,232	2,308	+3	Rikala and Jozefek 1990
<i>Paulownia</i>		6.5	5.5	16.3	17.6	+8	Turner et al. 1988
<i>Carya</i>		7.2	6.0	17,200	18,700	+9	Stafne and Carroll 2008
<i>Betula</i>		7.4	4.6	5,710	6,450	+13 (f)	Hoch 2018
<i>Koelreuteria</i>		6.0	5.0	1,200	1,400	+16	Wright et al. 1999a
<i>Carya</i>		6.5	5.5	13,380	15,990	+19	Sharpe and Marx 1986
<i>Quercus</i>	media	6.5	5.4	12,300	16,000	+30	Salifu et al. 2006
<i>Liquidambar</i>	GF endo	6.7	5.1	1,450	1,970	+36	Davis et al. 1993
<i>Quercus</i>	MM	6.1	4.5	2,600	3,600	+38	Browder et al. 2005
<i>Populus</i>	hybrid	6.5	4.0	3,800	5,400	+42 (f)	Böhlenius et al. 2016
<i>Robinia</i>		6.9	4.3	11,800	17,660	+49	McComb and Kapel 1942
<i>Fraxinus</i>		6.9	4.3	1,910	2,950	+54	McComb and Kapel 1942
<i>Populus</i>	1004	7.0	5.0	6,240	10,000	+60	DesRochers et al. 2003
<i>Salix</i>	GM endo	7.0	4.0	3,000	5,000	+66	Van der Heijden and Kuyper 2001
<i>Quercus</i>		7.4	4.0	6,000	10,000	+66 (f)	Beyer et al. 2013
<i>Quercus</i>		5.7	4.2	6,250	10,610	+69 (f)	Davis 2003
<i>Eucalyptus</i>	H495	6.6	4.6	3,000	5,300	+76	Aggangan and Malajczuk 1996
<i>Chestnut</i>	PT ecto	5.7	3.5	20,000	40,000	+100	Herendeen 2007
<i>Quercus</i>		7.4	4.6	2,810	6,510	+131 (f)	Hoch 2018
<i>Liquidambar</i>	GV endo	6.5	4.8	1,750	11,500	+557	Yawney et al. 1982

Supplemental fertilization [with nitrogen (N), phosphorus (P), and potassium (K)] was held constant in most of these trials, regardless of pH treatment. Better growth with increasing acidity to below pH 5.0 apparently results from higher nutrient use efficiency (e.g. uptake of N). For several species, the lower pH values increased biomass (and associated nutrients) by 15 percent or more.

In containers, the pH of the potting-mix will vary in both time and space. In some containers, the saturated area at the bottom of the container may be near at pH 7 while the top could be near pH 4 (Bumgarner et al. 2008). Also, the pH will change over time. In a yellow poplar study, pH declined from pH 5.3 to pH 4.1 in just 11 weeks

(Park et al. 2012). The amount of pH decline depends in part on how long the medium remains saturated (Ferrarezi and Testezlaf 2017).

4 Irrigation pH

In 2017, rain water in the eastern United States ranged from pH 4.9 to 5.7 while the range for irrigation water at bareroot nurseries is pH 4.5 to 10.1 (Figure 4). Although the pH reveals little about the quality of irrigation water, too much bicarbonates or too much sodium (Na) cause problems. For example, the Aliceville Nursery in Alabama closed due to irrigation water with high Na. The alkalinity of irrigation water can also affect the growth of container grown seedlings (Whitcomb 1988) as well as plants growing in sandy soils (Valdez-Aguilar et al. 2009). Iron (Fe) deficiency can occur when irrigation water contains high levels of bicarbonates (De LaGuardia and Alcántara 2002) and acidifying the water with phosphoric or sulfuric acid may improve growth. Even when bicarbonate levels are low, acidifying irrigation water with sulfuric acid (Miyamoto et al. 1975) can sometimes increase seedling growth. The growth response of 13 hardwood species to acidified water (pH 4 to pH 5) was positive (Table 6).

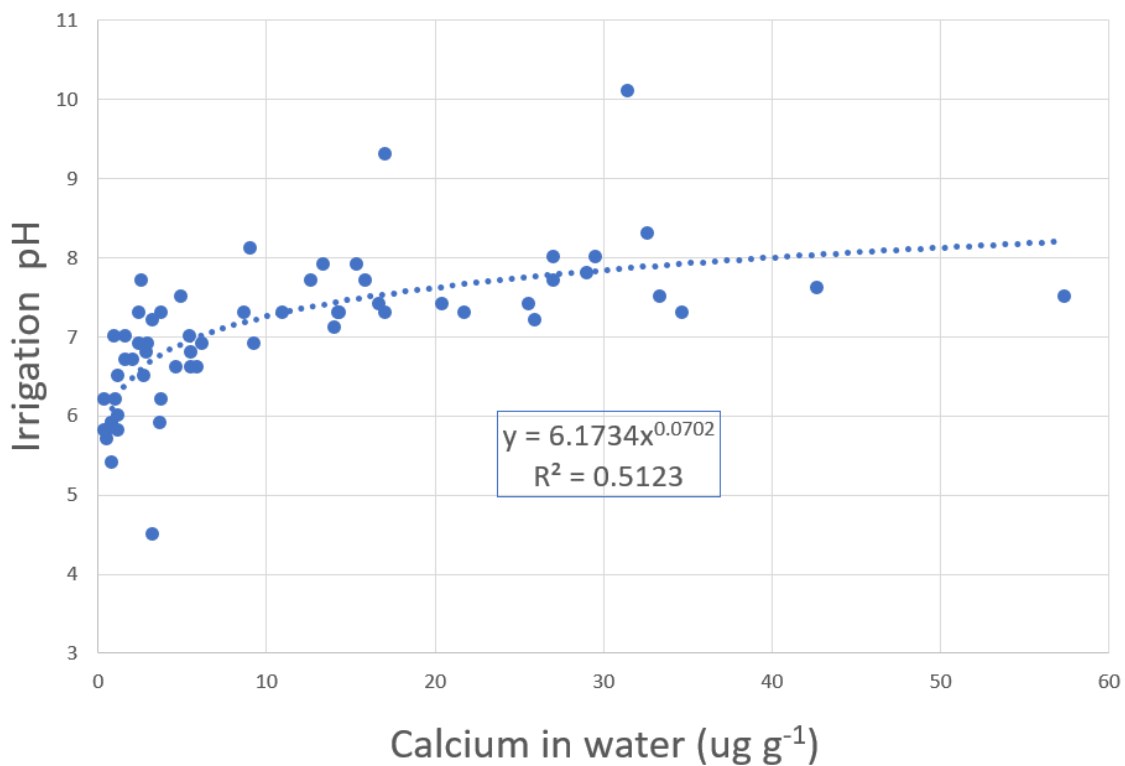


Figure 4. In general, the pH of irrigation water from nurseries in the South will be alkaline when it contains more than 10 $\mu\text{g g}^{-1}$ calcium. In contrast, water with less than 10 $\mu\text{g g}^{-1}$ calcium may range in pH from 4.5 to 8.1. Data from McNabb and Heidbreder-Olson (1998).

Some caution is recommended when making soil pH conclusions based on acid irrigation trials. Rainfall (less acidic than pH 4.0) typically does not injure foliage of hardwoods (Davis and Skelly 1992). In contrast, very acidic irrigation water (i.e. pH 3.0 to 3.5) can reduce growth due to foliar injury (Fan et al. 2000; Lee and Weber 1979; Lee et al. 1993; Miyamoto et al. 1975; Ramlall et al. 2015). One should not assume water that is artificially adjusted to pH 3.5 would produce similar results to growing seedlings in a soil at pH 3.5 (where foliar injury from acids does not occur). Likewise, one cannot assume that applying diluted hydrochloric acid to foliage (Hermle 2006; Long et al. 2017; Wingett 2005) will not injure hardwood seedlings. Although sulfuric acid can stimulate growth, acidifying irrigation water with hydrochloric acid may have no effect on growth (Browder 2004).

Table 6. Relative effects of acidifying irrigation water (nitric acid and/or sulfuric acid) on dry mass (mg) of plants or foliage (f). Mass #1 and mass #2 correspond to the dry mass of plants irrigated with water pH #1 or water pH #2, respectively.

Genus	Water pH #1	Water pH #2	Mass #1 (mg)	Mass #2 (mg)	Change in mass with decreased pH (%)	Reference
<i>Trichilia</i>	6.3	4.5	300	200	-33	Ramlall et al. 2015
<i>Koelreuteria</i>	6.0	3.5	8,170	7,090	-13	Fan et al. 2000
<i>Melia</i>	6.0	5.0	4,438	4,287	-3	Houbao and Chuanrong 1999
<i>Koelreuteria</i>	6.0	5.0	7,651	7,569	-1	Houbao and Chuanrong 1999
<i>Liriodendron</i>	5.5	3.0	9,220	9,220	0	Jensen and Patton 1990
<i>Betula</i>	5.6	3.5	1,090	1,120	+2	Kean and Manning 1988
<i>Quercus</i>	4.8	4.2	5,260	5,400	+3	Walker and McLaughlin 1991
<i>Acer</i>	6.0	4.5	108,000	111,000	+3	Medeiros et al. 2016
<i>Liriodendron</i>	5.6	4.3	2,590	2,670	+3	Chappelka et al. 1985
<i>Acer</i>	5.6	4.0	11,290	11,680	+3	Reich et al. 1986
<i>Betula</i>	5.6	3.5	760	790	+4	Keane and Manning 1988
<i>Quercus</i>	5.6	4.0	41,900	43,700	+4	Reich et al. 1986
<i>Acer</i>	5.2	4.0	429	460	+7	Raynal et al. 1982b
<i>Cinnamomum</i>	6.0	5.0	7,967	8,518	+7	Houbao and Chuanrong 1999
<i>Quercus</i>	6.0	4.5	40,000	45,000	+12	Medeiros et al. 2016
<i>Acer</i>	5.6	3.2	19,980	22,520	+13 (f)	Dixon and Kuja 1995
<i>Acer</i>	5.5	3.5	350	400	+14 (f)	Hogan 1998
<i>Fraxinus</i>	5.6	4.3	950	1,100	+15	Chappelka et al. 1986
<i>Michelia</i>	6.6	4.5	979	1,135	+16	Zongwei and Yunfeng 1991
<i>Cinnamomum</i>	6.0	3.5	7,990	9,610	+20	Fan et al. 2000
<i>Carya</i>	5.7	3.5	1,174	1,437	+22	Lee and Weber 1979
<i>Melia</i>	6.0	3.5	4,450	5,730	+28	Fan et al. 2000
<i>Ligustrum</i>	6.0	5.0	2,285	3,342	+46	Houbao and Chuanrong 1999
<i>Shorea</i>	6.6	4.5	1,288	1,932	+50	Zongwei and Yunfeng 1991
<i>Castanopsis</i>	6.0	5.0	1,211	2,536	+109	Houbao and Chuanrong 1999

5 Genetic differences

In the absence of properly designed nursery studies, people resort to speculating about the “optimum” pH range for growing hardwoods in bareroot nurseries. When experts resort to guessing, then it is likely that some pH ranges will conflict (Table 3). It is risky to make generic claims about the “optimum” soil pH range when using data from hydroponic studies or when evaluating just one genotype growing

in one soil type. Likewise, when comparing correlations between pH and seedling growth, presence of different mycorrhizal species may produce opposite results. The following examples illustrate some of the variability that exists among and within hardwood genotypes.

5.1 *Carya*

Wilde (1958) said pecan was an “exacting hardwood” that required a high level of nursery soil fertility with a pH range of 5.5 to 7.3. Although productive orchards are established on pH 8 soils (Sibbett 1995), Fe and zinc (Zn) may need to be applied to avoid a micronutrient deficiency. This indicates that when proper fertilizer management is used, the range of acceptable pH values can be increased. Sharpe and Marx (1986) grew pecan seedlings in pots and altered soil pH using $\text{Ca}(\text{OH})_2$. Although pH 6.5 was recommended, their single test indicated seedlings grew better at pH 5.5 (Table 5). When grown at pH 5.5, Zn fertilization was not required. In contrast, other researchers found that pecan grew better when soil pH was above 7.0 (Johnson and Hagler 1955; Sparks 1977). Could it be that ectomycorrhiza species (i.e. *Pisolithus tinctorius* vs. others) affect the direction of lime response in the same way as different endomycorrhiza affect sweetgum’s response to lime?

5.2 *Ilex*

Several varieties of Japanese holly are produced at bareroot ornamental nurseries. In one trial three varieties were grown in a sandy loam soil that was amended with dolomitic lime (Gilliam et al. 1985). One variety grew well at both pH 4.1 and 6.2 while two other varieties grew best at pH 4.1. The authors concluded: “these data show that in general, *Ilex* can be successfully grown at pH levels lower than the 5.5-6.5 range usually recommended.”

5.3 *Liquidambar*

Sweetgum can grow well at pH 4.5 (Ashby et al. 1989; Villarrubia 1980) and a few trials indicate that lime can reduce the growth of sweetgum (Figure 1; Yawney et al. 1982). Liming also reduced height growth when sterilized soil in a greenhouse was inoculated with *Glomus fasciculatum* (Davis et al. 1993). However, opposite results occurred when seedlings were inoculated with *Glomus mosseae* where the high rate of lime increased height growth by 3.6 cm. This suggests that soil biology is as important as tree species when determining the direction of response to lime.

5.4 *Paulownia*

Princess tree reportedly grows well in both acid and alkaline soils (pH 5-8.9) (Londo et al. 2006; Icka et al. 2016). Seedling growth in containers can be increased by adding NaOH to perlite (Melhuish et al. 1990). In Virginia, seedlings planted in pH 5.4-5.7 soil were over 1.2 m tall after two years in the field (Johnson et al. 2003). In greenhouse trials, the use of sulfuric acid on filter paper can reduce the germination of paulownia (Turner et al. 1988).

5.5 *Populus*

With proper fertilization, nursery managers can grow *Populus* in a wide range of soil pH, even in soils with a pH below 5 (Benzian 1965; Böhlenius et al. 2016; Hjelm and Rytter 2016). In fact, several researchers see no need to report pH values for either nursery or field trials with this species. This is likely due to the ease of growing stock in both acid (Davis et al. 2012; Hermle et al. 2006; Roberts and Gilliam 1995) and slightly alkaline soils (Carter 1981; Lombard et al. 2011; Puri et al. 2002; Shock et al. 2002). Although pH generalizations are often made for this genus (Table 3), variability exists among hybrids and species. Species like balsam poplar may not do well in “markedly” alkaline soils because Fe deficiencies may occur (Carter 1981; Jobling 1990; Lombard et al. 2010) while quaking aspen may die at pH 4.0 (Böhlenius et al. 2016). When certain hybrids were grown at low pH at a sandy nursery in the UK, growth was improved with copper (Cu) fertilization (Benzian 1965).

Although the pH range recommended for nurseries is pH 5.0 to 6.5 (Jobling 1990), the range for plantations can include alkaline soils (Baker and Broadfoot 1979; Shock et al. 2002). The belief that well-fertilized cuttings do not grow well at in plantations at pH 3.5 to 4.0 might be due to greenhouse trials where cuttings died (Böhlenius et al. 2016). Regardless of soil acidity, proper fertilization is important. Growth of cuttings can be reduced when too much P is applied to nursery soil at pH 6.2 (Teng and Timmer 1990).

5.6 *Quercus*

Several liming studies with northern red oak have been conducted in greenhouses. Phares (1964) added lime (CaCO_3) to two soil types and repeated the study over two years. He detected a year x lime x soil type interaction. For one soil type, adding lime reduced growth in both years but for another soil type, liming increased growth in one year and decreased growth in another. The three-way interaction points out the danger of making too broad a generalization based on a single pH trial. In some cases, applying lime may correct a deficiency in Ca while in other cases, it might induce a P deficiency. Overall, adding $1,120 \text{ kg ha}^{-1}$ of lime to soil (pH 5.8) reduced heights and biomass by 10 and 4 percent, respectively. Phares concluded that increasing soil pH with lime caused problems with the uptake of P.

When northern red oak seedlings were grown in pots in a greenhouse, lime (Ca(OH)_2) did not improve growth in six trials but growth was reduced in one trial (Joslin and Wolfe 1989). Beyer et al. (2013) also found that too much lime (CaCO_3) reduced growth of northern red oak growth (Figure 3; Table 5). Although Fe chlorosis may occur when oak is planted in alkaline soils (Hacskeylo and Struthers 1959; Thomas et al. 1998), not all chlorotic oaks are Fe deficient (Watson and Himelick 2004).

Fertilized oak seedlings grow well in soils below pH 4 (Beyer et al. 2013; Davis 2003) and in media at pH 4.5 (Olchowik et al. 2017). Likewise, oak plantations also perform well at pH 4.5 (Emerson et al. 2009). Fertilized oak seedlings can grow well at pH 3.7 (Hanson et al. 1986), pH 5.8 (Hart and Sharp 1997) or at pH 7.2 (Schmal et al. 2011). In contrast, some believe that sawtooth oak should not be planted on strip mine sites with soil pH <4.0 (Kennedy and Krinard 1985).

6 Warnings about low soil pH

Experiences with growing tree seedlings (Tables 4 and 5) lend support to “the fact that the toxicity of hydrogen ions has been grossly exaggerated” (Wilde 1954). The exaggerated claims originated from “artificially prepared cultures,” not soil studies (Kidd 2001; Wilde 1954). In most acid soils aluminum (Al) and manganese (Mn) toxicities are probably more important than the amount of hydrogen ions (Adams and Foy 1984; Pope 1984). Although some claim soil lower than pH 5.5 is not optimum for growth of fertilized hardwoods (Table 1), most provide no data to show their warnings have merit. Some admit they do not know what problems might result when adequate fertilizers are applied to low pH soils (Stone 1965). Others believe that Al toxicity “may” occur when soil pH is low (Sparkes 1977). Although Al toxicity symptoms can occur when soil contains $101 \mu\text{g g}^{-1}$ (Pope 1984), no “toxicity” occurred at pH 3.3 when soil contained $44 \mu\text{g g}^{-1}$ of Al (Figure 3). There are no reports from sandy nursery trials to show Al levels are high enough to cause any problems with hardwoods. Although northern red oak is supposed to be sensitive to Al (DeWald et al. 1990, Kelly et al. 1990), seedlings grew well in soil with more than $700 \mu\text{g g}^{-1}$ (Lesko and Jacobs 2018). In one greenhouse trial, oak seedlings grew better when the potting-mix was pre-treated with aluminum sulfate (Davis 2003).

At bareroot hardwood nurseries (Table 2), micronutrient toxicities are rare in very acid soils (pH 4.0-5.0). Some wonder if poor growth of hardwoods might occur on low pH soils (with low cation exchange capacity) due to deficiencies of K, Ca and Mg. Regardless of soil pH, poor growth can occur in sandy soils when fertilization is not adequate. In sandy soils, seedling growth is positively related to the amount of soil cations (Ouimet et al. 1996; South et al. 2018). For this reason, managers typically apply fertilizers to increase growth of hardwoods in low pH soils.

Zn toxicity can occur at locations where Zn has contaminated the soil (Denny and Wilkins 1987; Shi et al. 2016) but at nurseries it is unlikely since sandy soils typically contain less than $12 \mu\text{g g}^{-1}$ Zn (South and Davey 1983). Although Zn levels of $50 \mu\text{g g}^{-1}$ (Mehlich III) might reduce growth of sensitive species like northern red oak (Beyer et al. 2013), other oaks are tolerant of soil levels as high as $1,600 \mu\text{g g}^{-1}$ (Shi et al. 2016).

High levels of Mn in fine-textured nursery soils (pH<5) can reduce growth of hardwood (Child and Smith 1960; Wong 2005) and pine seedlings (South 2017). Growth can be negatively related to Mn levels when seedlings are grown in water (Kitao and Koike 1999; Thomas and Sprenger 2008) or soil (Böhlenius et al. 2016; Lei et al. 2007). Although high levels of Mn in irrigation water can reduce growth of sugar maple (Schier and McQuattie 2000; St.Clair and Lynch 2005), Mn is not considered a problem in most irrigation waters (Landis 1989). Even runoff containment basins at container nurseries typically contains less than 0.4 mg l^{-1} (Copes et al. 2017). In fact, there are no reports of cases where Mn toxicity has occurred in bareroot hardwood nurseries where soil contains less than $300 \mu\text{g g}^{-1}$ Mn. Most sandy nursery soils contain less than $200 \mu\text{g g}^{-1}$ Mn.

7 Advantages of low pH

In the past, lowering soil pH with sulfuric acid was an effective pest management practice (Bickelhaupt 1987; Hartley 1921; Jackson 1933; Wilde 1937). When soil acidity increases, beneficial fungi like *Trichoderma* and *Penicillium* may also increase (Huang and Kuhlman 1991). Populations of damping-off fungi, *Fusarium*,

Phytophthora, nematodes, and certain weeds (Aldhous 1972; Buchanan et al. 1975; Goswami et al. 2011; Huang and Kuhlman 1991; James 1997; Khilare and Ahmed 2012; Shareef et al. 2016; Stoeckeler and Jones 1957) may all be lower when nonfumigated seedbeds have pH values less than 5.0. Damping-off of hardwoods may be less when soil pH is below 6.0 (Enebak 2019). It has been suspected for some time that nematode populations are lower in acid soils (Wilde 1934) and several studies support this view (Burns 1971; Korthals et al. 1996; Willis 1972).

Although data on the relationship between soil pH and damping-off of hardwoods are limited, some researchers recommend pH be kept below 6.1 (Hodges 1962; Hooper and Curtin 1983; Lamichhane et al. 2017). For example, when soil pH was 6.0-7.8, damping-off of hardwoods was high at three silty-clay-loam nurseries (Wright 1944). Fortunately, a few greenhouse trials report the effects of irrigation acidification on seed germination. When compared to pH 5.6-5.7, germination of yellow birch was as good or better when seeds were irrigated with pH 3.5-3.6 water (Lee and Weber 1979; Percy 1986). This is not unexpected since pre-treating seed with sulfuric acid can increase germination of some hardwoods (Li et al. 2013; Peacock and Hummer 1996). In contrast, the germination rate of ten other hardwood species was not affected by pH 4 water (Fan et al. 2000; Lee and Weber 1979) while germination of red maple was 3 percent lower (Raynal et al. 1982a).

8 Sulfur

Sulfur (S) is used to lower soil pH in bareroot nurseries (South 2017) and when growing maple, some apply elemental S when soil pH is greater than 5.6 (Altland 2006). For other hardwoods, S is applied when soil values exceed pH 6.0-6.2 (Davey and McNabb 2019; Messenger 1984). For example, at one nursery with pH 6.0 soil, adding 374 kg ha⁻¹ of S increased height and diameter growth of sweetgum and green ash (Stone 1980; Villarrubia 1980).

Although peat media are typically acidic, S or sulfuric acid is sometimes added to reduce media pH (Wightman et al. 2001; Will and Faust 1999). In one study, applying sulfuric acid doubled the size of oak seedlings (Browder 2004). Sometimes S can improve growth even when a change in pH is not needed. Applying solutions (containing KCl and K₂SO₄) to pH 4.1 pine bark medium demonstrated that growth of oak and maple in containers was related to the amount of sulfur added (Browder et al. 2005). Likewise, 100 kg ha⁻¹ of S increased the growth of container-grown quaking aspen growing in pH 5.5 soil (Liang and Chang 2004).

Applying S just before sowing can reduce growth under certain weather conditions (Aldhous 1972; South 2017). Therefore, to reduce the risk of root injury, sulfur should be applied at least two to three months before sowing (Armson and Sadreika 1979). Applying too much S just before transplanting can reduce root growth of poplar cuttings (Timmer 1985), and applying too much aluminum sulfate before sowing can reduce germination of different species (Table 4), especially when seedbeds are not kept moist (Russell 1990). Some experts do not recommend applying aluminum sulfate due to potential toxic effects (Iyer et al. 1969; Pope 1982; Voigt 1954; Whitcomb 2009). Except for maple, there is typically no need to add S when hardwood nursery soils range from pH 4.5 to 6.5 (unless there is a risk of a S deficiency which limits growth).

9 Lime

“The old concept concerning lime was that of a *cure-all* - that it was almost certain to be beneficial, no matter what the problem. Such an opinion should now be discarded since it may lead to a waste of money and in some cases to over liming. The use of lime must be based on measured soil acidity and on crop requirements” (Brady 1974).

Adding lime to soil (Table 4; Crews and Dick 1998; Gabriel et al. 2018; Wilde 1958) or potting media (Altland and Jeong 2016; Hjelm and Rytter 2016) will sometimes increase growth of hardwoods. However, it can be a waste of money when there is no effect on growth (Figure 1; Böhlenius et al. 2016; Crannell et al. 1994; Hannah 1973) or when growth is reduced. In greenhouse studies, adding lime had negative effects on choke cherry, river birch (Dumroese et al. 1990), eucalyptus (Handreck 1989; Symonds et al. 2001), oak (Browder et al. 2005; Denig et al. 2014; Hoch 2018; Wright et al. 1999b), blackgum (Wright et al. 1999b), *Prosopis* (Cline et al. 1986), and *Populus* (Timmer 1985). Some believe a reduction in growth is not due to an increase in alkalinity but to an increase in demand for nitrogen by saprophytic bacteria (Wilde 1946). In addition, root injury may occur when calcium hydroxide (slacked lime) is applied before sowing (Wilde 1958). As a precaution, Turner (1970) recommends applying lime prior to sowing a cover crop (i.e. a year before sowing hardwoods). Adding too much lime will sometimes induce deficiencies in Cu (Anderssen 1932), Fe (Hacsckaylo and Struthers 1959; Handreck 1989; Hoch 2018), P (Cline et al. 1986) and Mn (Altland 2006; Messenger and Hruby 1990). For these reasons, nursery managers in the southern United States typically do not lime seedbeds until the pH falls below pH 4.8.

10 Problems with some conclusions

With few exceptions, there is always some confounding when conducting studies with S, lime, sulfuric acid and hydrochloric acid. As a result, conclusions about the “optimum” pH for most trials should be considered “tentative”. For example, if species do not grow well in the presence of chloride ions, then adding hydrochloric acid to perlite may explain why pH was positively related to seedling growth (Melhuish et al. 1990). Is it possible the hydrogen ion concentration was not the primary reason for poor root growth (Whitcomb 1983; Wilde 1954)? When positive responses are observed after applications of dolomitic lime (e.g. Crews and Dick 1998; Voigt et al. 1958), could this be due to correcting deficiencies in soil Ca, or Mg, or micronutrients?

Some researchers study pH effects by comparing different types of soils (Davidson 1979; Hjelm and Rytter 2016). In one trial, soil from two strip-mined sites (pH 4.5 and 5.5) was compared to a commercial loam soil at pH 6.5 (Turner et al. 1988). Was the better growth on the 6.5 pH loam due to less acidity or was it due to beneficial variables associated with the commercial product?

Some trials are designed to have a minimum amount of confounding. One well designed study examined the effects of S on seedling growth (Browder 2004). Instead of applying S to the potting-mix (which affects pH), the authors applied various rates of K_2SO_4 and kept K levels constant by adding KCl. Although this added confounding with Cl, their approach removed pH as a confounding factor.

To improve interpretation of results, gypsum treatments should be included with lime ($CaCO_3$) treatments (Figure 5). That will allow researchers to compare the relative growth responses to equal rates of Ca (Bakker et al. 1999; Deines 1973; Fuller

and Meadows 1983; Gabriel et al. 2018; Marx 1990). As a result, conclusions (about the reasons for pH related growth responses) can be made with greater confidence.

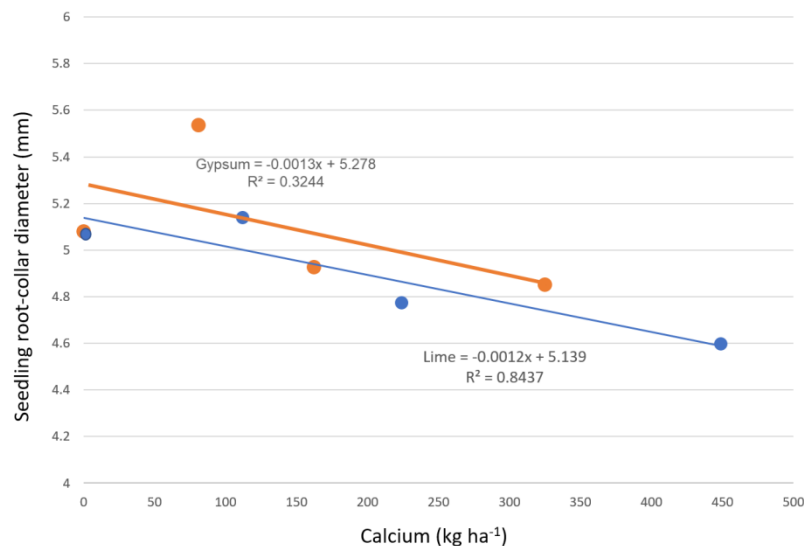


Figure 5. The effect of adding calcium on diameter growth of sweetgum seedlings at the Edwards Nursery in North Carolina (Deines 1973). The linear relationship between calcium rate and diameter growth is approximately the same for both lime (CaCO_3) and gypsum (CaSO_4). At this nursery, soil pH is apparently not the driving factor behind a significant reduction in diameter growth.

11 Future research

Greenhouse and nursery researchers should question traditional liming and pH recommendations (Gilliam et al. 1985; Ruter and van der Werken 1986; Whitcomb 1983) and should target pH values that are near optimum for the species under investigation. Unfortunately, for many tree species well-designed “optimal pH” trials have not been published. Therefore, “Assessment of a desirable pH range of a given species” should be one of the first factors investigated (Bryan et al. 1989). For a given environment, researchers should determine a peak pH value where seedling growth is maximized (Figure 2). There could be several ways to identify this peak, but one method might involve adding eight rates of lime to a low pH soil (e.g. Handreck 1989) and comparing the results from a “twin” study where the same pH levels were achieved by applying sulfuric acid (several months before sowing) to a high pH soil.

Researchers conducting bareroot trials can test several pH related hypotheses. One testable hypothesis is that micronutrient deficiencies do not occur when growing fertilized 1-0 hardwood seedlings at pH 4.5. Although micronutrient deficiencies can occur in alkaline nursery soils, there are only a few published photos of bareroot hardwoods, with micronutrient deficiencies, in seedbeds at pH 4.5 (e.g. Benzian 1965). Most micronutrient deficiency photos are taken after hardwood seedlings are grown in hydroponics or nutrient-deficient sand (e.g. Erdmann et al. 1979; Hacskaylo et al. 1969; Whittier 2018).

Several greenhouse trials suggest that nutrient use efficiency is increased when soil acidity increases (Table 5). Although the reason is not clear, some studies indicate that acidification could increase the uptake of nitrate (Paez-Valencia et al. 2013). Research aimed at a better understanding of why low pH increases uptake of nutrients should be initiated. One testable hypothesis is: for nutrient use efficiency (N), there is no species by pH interaction.

Too much elemental S applied immediately before sowing (e.g. 900 kg ha⁻¹) can sometimes reduce seedling growth (but not always). To reduce the risk of injury from the formation of sulfuric acid, elemental S should not be applied one or two months before sowing. Future research should be conducted to better quantify the factors that contribute to injury from exposure to sulfuric acid that forms after fertilization with elemental S.

12 Conclusions

Field observations and greenhouse trials confirm that pH 6.0 to 8.0 is not optimum for growing most hardwoods in either bareroot or container nurseries. Therefore, managers of sandy, bareroot nurseries grow both hardwoods and pines in soils that range from pH 4.8 to 6.0. Based on published data, the desired range for growing oak seedlings at bareroot nurseries (with more than 75 percent sand) is likely pH 4.5 to 6.0.

13 Acknowledgments

The author thanks Jake Stone who was the first to inform me that “very strongly acid” soils (pH 4.5) were not harmful when growing green ash and sweetgum seedlings at his nursery. The author thanks J.B. Jett (North Carolina State University), John Mexal (New Mexico State University), Robert Cross (Cross Consultants), Chase Weatherly (Arborgen), Carl Whitcomb (Lacebark Research) and Nicky Jones (Sappi) for providing helpful comments on the manuscript.

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